

## ANTIBACTERIAL COATING FOR SHOE INSOLES WITH TITANIUM DIOXIDE NANOPARTICLES

DEISLAVA S. ANGELOVA<sup>1</sup>, DARINA ZHELEVA<sup>1</sup>,  
 YANA GOICHEVA<sup>2</sup>, EKATERINA KRUMOVA<sup>2</sup>

<sup>1</sup>*Department of Textile, Leather and Fuels, University of Chemical Technology and Metallurgy, Sofia, Bulgaria,  
 E-mail: [darinajeleva@abv.bg](mailto:darinajeleva@abv.bg)*

<sup>2</sup>*Stephan Angeloff Institute of Microbiology, Sofia, Bulgaria*

To protect the human foot from the attack of bacteria and fungi, it is extremely important to choose an appropriate antimicrobial agent. The main component building skin tissue is collagen. Hydrolyzed collagen is widely used in medicine due to its properties of biodegradability, biocompatibility and non-toxicity. It is characterized by excellent film-forming properties. TiO<sub>2</sub> nanostructures have been widely studied as antimicrobial agents due to their photocatalytic activity under UV light and possess excellent antibacterial properties. The application of nanomaterials to treated leather surfaces resists the growth of microorganisms as well as exhibiting a self-cleaning effect. In this study, TiO<sub>2</sub> nanoparticles were *in situ* synthesized using oxalic acid as a precursor. These particles were deposited on the cross-linked gelatin hydrogel as finishing layer applied to shoe insoles. SEM, UV-Vis, FTIR and antibacterial tests for Gram-positive and Gram-negative bacteria were performed. Microscopic observations prove a distribution of titanium dioxide nanoparticles and enveloped by the collagen film. The presence of chemical and physical bonds between the different components of the biocomposite film has been demonstrated. The antimicrobial activity of the investigated shoe insole samples was evaluated by the reduction of bacterial growth. Shoe insoles coated with gelatin-titanium dioxide nanocomposite show high antibacterial activity against the strains used. Consequently, incorporating metal particles such as TiO<sub>2</sub> into the cross-linked collagen hydrogel and fixing it to the treated leather shoe material is a good alternative for antimicrobial treatment.

Keywords: antibacterial finishing, insole, TiO<sub>2</sub> nanoparticles

### INTRODUCTION

Natural leather is the most commonly used material for shoe insoles. Because of its absorbent and breathable properties, leather has unique characteristics, climate-balancing, and good hygienic effects on the shoe. The main component of leather is protein; therefore, it creates conditions for breeding microorganisms, including bacteria and fungi during the storage and use process, which may affect the structure and properties of leather and damage the health of users. Moreover, the human foot is in a continuous condition of increased temperature, pH and humidity. High levels of hydration, humidity and ambient temperatures, lead to an increase in the number of microorganisms, especially *Staphylococcus aureus* (Healy *et al.*, 2010). Commonly used protection chemicals are mostly volatile organic compounds, due to their carcinogenic effect and environmental toxicity, it is prone to become unacceptable (Bai *et al.*, 2022). This has led to the idea of developing antimicrobial material leather for insoles with a more durable and effective treatment without the use of harmful chemicals.

Considerable research has been focused on the treatment of leather with nanometal oxides during the pre-tanning process, where nanomaterials are added to the tanning solution. However, there are still few studies that deal with the nanoprocessing of leather in the final finishing step without adding the nanomaterial to the tanning solution. Several nano-metal oxides (Ag<sub>2</sub>O, ZnO, MnO<sub>2</sub>, CuO) have great, causing destruction of the cell wall and

considerable binding affinity toward thiol -SH- containing groups, showing significant antimicrobial effects (Muthukrishnan *et al.*, 2019; Wang *et al.*, 2017).

A sole leather loaded with nano-zinc oxide has been developed and the treated leather samples showed significant resistance against fungus and bacteria (Habib *et al.*, 2023). Titanium dioxide nanoparticles (TiO<sub>2</sub>-NPs) are already used in various practical applications, such as water and air purification, self-cleaning and self-sterilizing surfaces, and optical and dielectric devices. One of TiO<sub>2</sub> uses is as a pigment because of its brightness, high refractive index, and resistance to discoloration. Because of its interesting photocatalytic properties, TiO<sub>2</sub> has been used in decontamination, purification, and deodorization of air and wastewater but it also has the ability to kill cancer cells, bacteria and viruses under mild UV illumination. TiO<sub>2</sub> proves to be the most suitable safe and broad-spectrum antimicrobial agent (Sunada *et al.*, 2003). Studies have shown an increase in the antimicrobial activity of leather coated with Ag-TiO<sub>2</sub> NPs (Carvalho *et al.*, 2018; Khalid, 2022; Gaidău *et al.*, 2016; Kaygusuz *et al.*, 2016; Marques *et al.*, 2022; Ignat *et al.*, 2020). Others are considering the possibility of use in the leather treatment process with the addition of the TiO<sub>2</sub>-SiO<sub>2</sub> nanocomposite (Kaygusuz *et al.*, 2017).

The finishing film of leather can vary greatly, mainly depending on its purpose. The sol-gel method is one of the most widely used techniques for the synthesis of functional coating films for substrate surface modification and the improvement of material properties which are affected by surface conditions (Tan *et al.*, 2021). This technique possesses a number of advantages over conventional film formation techniques, including a relatively low processing temperature, ease of applying homogeneous multicomponent oxide films over large surfaces, and good control of the composition and properties of the final material.

In this study, a finish coating was successfully obtained by modifying leather samples with cross-linked gelatin containing TiO<sub>2</sub> particles, which were synthesized *in situ*.

## MATERIALS AND METHODS

### Materials

Natural pig leather from a local producer, chrome-tanned, no finish, 0.89 mm thick. The test pieces are 100/50 mm in size, and the average weight of the samples is 2.4 g. TiO<sub>2</sub> from Sigma-Aldrich (Darmstadt, Germany); Glutaraldehyde (25% aqueous solution) from Sigma-Aldrich (Darmstadt, Germany); Gelatin from Merck KGaA (Darmstadt, Germany); and oxalic acid (H<sub>2</sub>C<sub>2</sub>O<sub>4</sub>) from Merck KGaA (Darmstadt, Germany).

Three variants of modification of leather materials were applied: *LT-1* (Le-TiO<sub>2\_1</sub>), *LT-2* (Le-TiO<sub>2\_2</sub>), and *LT-3* (Le-TiO<sub>2\_3</sub>). The finishing films were obtained by immersion in the different solutions in equal concentrations. For all samples, aqueous solutions were used in the following concentrations: 5% gelatin solution, 2.5% GA (glutaraldehyde), 0.1 M TiO<sub>2</sub>, and 0.1 M C<sub>2</sub>H<sub>2</sub>O<sub>4</sub>.

#### *LT-1 (Le-TiO<sub>2\_1</sub>)*

The leather samples were immersed in a solution of 0.1 M TiO<sub>2</sub> and a 0.1 M C<sub>2</sub>H<sub>2</sub>O<sub>4</sub> for 30 min at t=60°C. TiO<sub>2</sub> particles were synthesized *in situ*. After that, the samples removed from the solution and immediately immersed in solution of gelatin and finally the glutaraldehyde solution was added for crosslinking. The leather samples were dried for 2 h at t=60°C. The samples were rinsed with distilled water and left to dry at room temperature.

### LT-2 (*Le-TiO<sub>2</sub>\_2*)

The leather samples were immersed in a gelatin solution for 2 h at  $t=60$  °C. After that, the samples were immersed in the solution of TiO<sub>2</sub> and oxalic acid for 30 min at  $t=60$ °C. The next stage was adding the glutaraldehyde. Finally, the samples were washed and dried.

### LT-3 (*Le-TiO<sub>2</sub>\_3*)

The leather samples were submerged into previously prepared solution of gelatin and glutaraldehyde. After crosslinked samples were immersed in solution of TiO<sub>2</sub> for 30 min and heated at  $t=60$ °C. The next stage was immersion in a solution of oxalic acid for 30 min. Then the samples are rinsed with distilled water and dried at 60 °C for 2 h.

## Analysis

The surface morphology of the modified materials and the formation of TiO<sub>2</sub> particles were analyzed using a scanning electron microscope (SEM) Philips ESEM XL30 FEG. FTIR analysis was performed on a Fourier transform infrared spectrometer (IRAffinity-1, Shimadzu, Japan), the spectral range of 4000÷600 cm<sup>-1</sup>. Spectral characteristics are taken using UVA/VIS/NIR spectrophotometer Lambda 750S PerkinElmer, USA, in range of length  $\lambda=2500\div250$  nm. The leather physical and mechanical tests: Determination of water vapour permeability (ISO 14268:2023) was carried out with apparatus SATRA STM 473. The water vapour absorption indicator is determined by calculation (ISO 17229:2016).

The antimicrobial activity of the tested leather samples was evaluated using the Kirby–Bauer disc diffusion method (Hudzicki, 2009). The test bacterial strains used in this study included *Pseudomonas aeruginosa* 1390 (Gram-), *Bacillus subtilis* 168 (Gram-) and *Escherichia coli* W 1655 (Gram-): sourced from the National Bank of Industrial Microorganisms and Cell Cultures, Bulgaria, along with *Erysipelothrix rhusiopathiae* B40 (Gram+) from the culture collection of The Stephan Angeloff Institute of Microbiology. Bacterial cultures were first grown overnight in Nutrient Broth (HIMedia, India) at 37 °C, and the cell density was then standardized to McFarland 0.5. Sterile Mueller-Hinton agar plates were inoculated with the standardized bacterial suspensions. Antibiotics known to be effective against the test microorganisms were used as a positive control. The antimicrobial activity was determined after 24 h of incubation by measuring the diameter (mm) of the inhibition zones (ZOI) around each disc.

## RESULTS AND DISCUSSION

### Morphological Properties of Modified Leather Samples with TiO<sub>2</sub>

The SEM micrographs in Fig. 1 show the surface of a treated and untreated leather samples. It can be seen that the control leather sample (C-control) has the characteristic fibrillar collagen structure. TiO<sub>2</sub> particles appear clearly on the treated surface: LT-1, LT-2, LT-3, indicating the success at their deposition. In addition, the pores are clearly visible on the surface of treated leather, suggesting that the surface coating is thin. Microscopic studies showed that TiO<sub>2</sub> particles were impregnated into the structure of the gelatin hydrogel at LT-1 and LT-2, and were distributed into small spherical particles, while in LT-3 the particles are on the surface of the film. TiO<sub>2</sub>-NPs has polyhedral shape with rounded edges. It is worth mentioning that small bright spots can also be identified in LT-1 and LT-2 images, which probably are non-agglomerated. Greater aggregations were observed on the leather surface in LT-3, compared to the other variants studied.

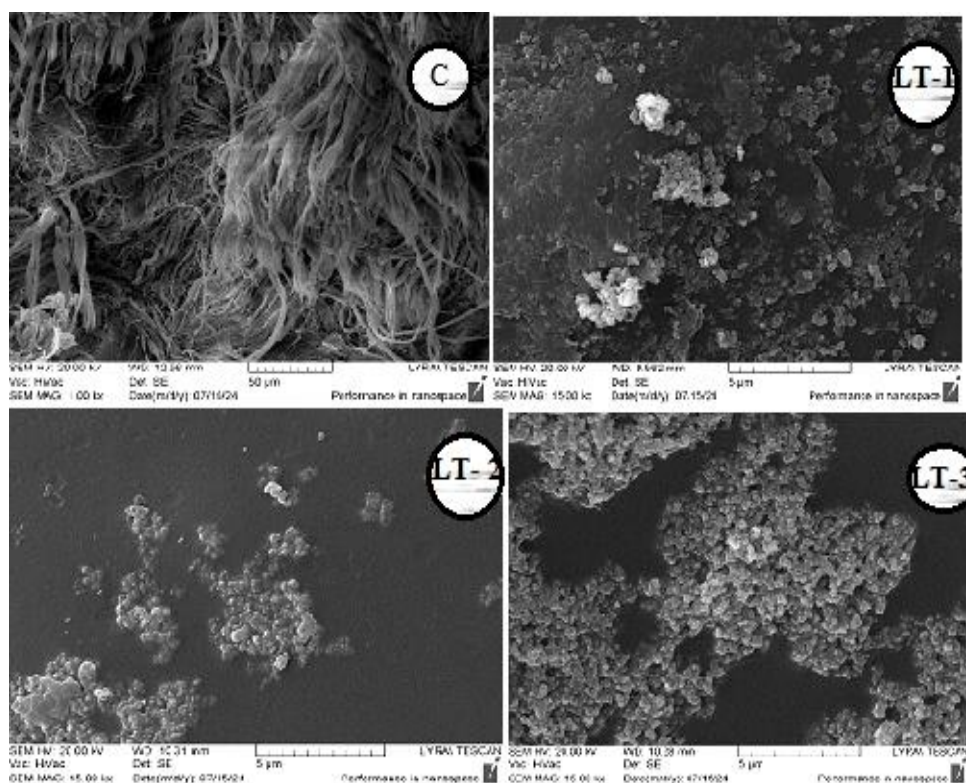


Figure 1. SEM images of the surface of the leather: C – control sample; LT-1, LT-2, LT-3 are modified leather samples

### Determination of Water Vapour Permeability and Water Vapour Absorption

Table 1 shows how the various treatment options affect water vapour absorption and permeability. The samples LT-2 have reduced vapour permeability and water vapour adsorption, the samples LT-1 have increased vapour permeability and water vapour absorption as a result of the coating. The LT-3 provided the highest results, consequently these materials have the best absorbent properties and climate-balancing impact for shoe insoles.

Table 1. Vapour permeability and absorption tests of leather samples

Sample	Water vapour absorption $W_1$ $mg/cm^2$	Water vapour permeability $W_3$ $mg/(cm^2.h)$	Vapour permeability coefficient $W_2$ $mg/cm^2$
Le (Control)	2.17	17.41	141.45
LT-1	2.89	16.12	131.85
LT-2	1.47	13.20	107.07
LT-3	3.13	18.42	150.49

### FTIR and UV-VIS Analysis

Fig. 2 represents the UV-VIS spectra of  $TiO_2$  that showed the maximum absorption peak at 365 nm, which was a primary sign of the successful formation of  $TiO_2$ -NPs (Anandgaonker *et al.*, 2019).



One possible explanation of the modified samples bactericidal effect is that it is due to the presence of Ti particles with a positive charge of titanium ions (Ti<sup>+</sup>) on the surface of the leather, which can cause oxidative stress and destruction of the bacterial cell wall.

## CONCLUSIONS

A finish coating was successfully obtained by modifying leather samples with cross-linked gelatin containing TiO<sub>2</sub> particles, which particles were synthesized *in situ*. Three variants of synthesis of TiO<sub>2</sub> NPs were investigated. It was proved that LT-2 shows significant resistance against 4 types of bacteria. Consequently, these finishes can be very effectively used as protective antibacterial coatings for shoe leather materials, protecting the human foot from the effects of microorganisms.

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